

## Ammonium oxalate monohydrate Data Sheet

| Catalog sizes         | <ul><li>100g, Listed as 100g Ammonium oxalate monohydrate, AMOX00100</li><li>500g, Listed as 500g Ammonium oxalate monohydrate, AMOX00500</li><li>1kg, Listed as 1kg Ammonium oxalate monohydrate, AMOX01000</li></ul>  |
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| Category              | Ionic Liquids   |
| Product specification | <ul> <li>White crystalline powder</li> <li>Product ID : AMOX00100</li> <li>Purity : 98%+</li> <li>CAS : 6009-70-7</li> <li>Formula : C2H8N2O4</li> <li>MW : 142.11g/mol</li> <li>MP : 131-135C</li> <li>Solubility : 40-50g/L in water</li> </ul>   |
| Product description   | <ul> <li>Ammonium oxalate monohydrate is a white, odorless, crystalline,<br/>hygroscopic, water-soluble salt. One principal use of ammonium oxalate is<br/>in the controlled production of calcium oxalate to generate an acid<br/>resistant, protective coating for construction stones (limestone, marble,<br/>and calcareous sandstone artefacts) and plaster [Mundronja D. Applied<br/>Physics (2013) 111: 109-119]. The displacement reaction between<br/>ammonium oxalate and calcium carbonate to form calcium oxalate is the<br/>premise on which this treatment is based. Ammonium oxalate is also used<br/>in to manufacture explosives, industrial dyes, buffering agents, and<br/>polishing substances. Such uses can be generally summarized as: <ul> <li>Decalcification agent</li> <li>Surface treating agent for limestone, marble, and calcareous<br/>sandstones</li> <li>Ferrous iron extraction agent</li> <li>Protection of monumental limestone</li> <li>Anticoagulant</li> <li>Analytical reagent</li> <li>Reducing agent</li> <li>Precipitant</li> </ul> </li> </ul> |